

1.

①

$$a) K_p = \frac{P_{H_2O}^2}{P_{SO_2} \cdot P_{H_2S}^2} \quad 1pt$$

/1pt

$$b) \Delta_r G^\ominus = -RT \ln K_p = \Delta_r H^\ominus - T \cdot \Delta_r S^\ominus \quad | : -RT \quad 1pt$$

$$\ln K_p = \frac{\Delta_r S^\ominus}{R} - \frac{\Delta_r H^\ominus}{RT}$$

/2pt

c) grafiks pierienots:

C → K → 1/K - 2pt

K → ln K → 1pt

$$\text{virtuena koef.} : - \frac{\Delta_r H^\ominus}{R} = 12\,665 \quad \text{grafiks} - 2pt$$

$$\begin{aligned} \Rightarrow \Delta_r H^\ominus &= -12\,665 \times 8.3145 \\ &= \underline{\underline{-105\,125 \text{ J} \cdot \text{mol}^{-1}}} \quad 2pt \end{aligned}$$

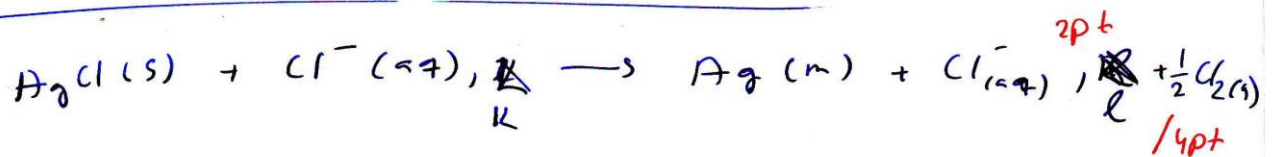
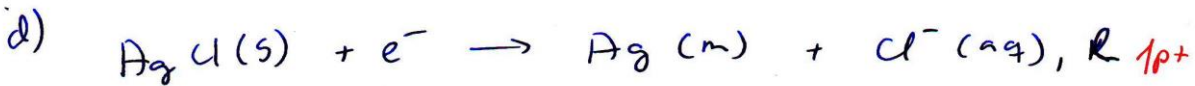
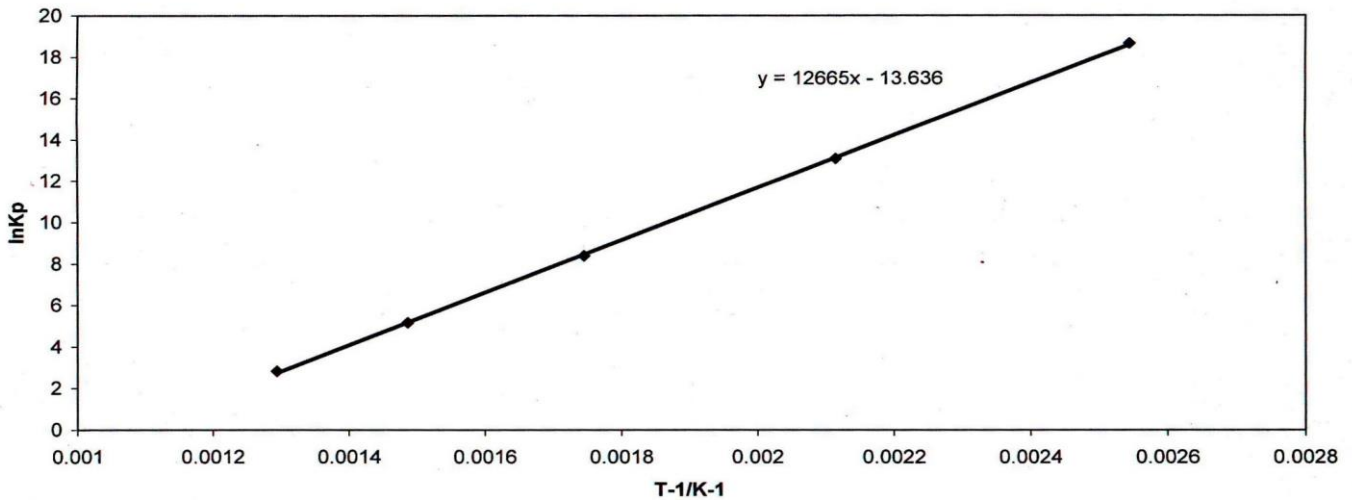
$$\text{krustpunkts ar y asi: } \frac{\Delta_r S^\ominus}{R} = -13.636$$

$$\begin{aligned} \Delta_r S^\ominus &= 8.3145 \times 13.636 \quad 2pt \\ &= \underline{\underline{-113 \text{ J} \cdot \text{mol}^{-1} \cdot \text{K}^{-1}}} \end{aligned}$$

/9pt

T/C	T/K	T-1/K-1	Kp	ln(Kp)
120	393	0.002545	1.26E+08	18.65179
200	473	0.002114	4.84E+05	13.08984
300	573	0.001745	4.41E+03	8.39163
400	673	0.001486	1.75E+02	5.164786
500	773	0.001294	1.69E+01	2.827314

Grafiks entropijas un entalpijas noteikšanai

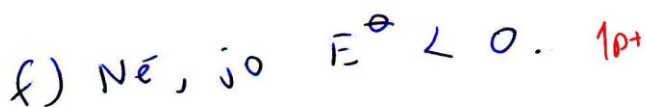


e) $E = E^\ominus - \frac{RT}{F} \ln \frac{\sqrt{P_{\text{Cl}_2}} \cdot [\text{Cl}^-]}{[\text{Cl}^-]}$ 1pt

$\rightarrow E^\ominus = E + \frac{RT}{F} \cdot \ln \dots$

$= -1.1363 + \frac{8.314 \times 293}{96500} \cdot \ln \frac{10^{-3}}{10^{-4}}$ 2pt

$= \underline{\underline{-1.1372 \text{ V}}}$ /3pt



/1pt

20pt
kopā.

$$a) r_i = \underline{2f k_d [I]}$$

3pt

$$b) \underline{[I]_t = [I]_0 e^{-k_d t}}$$

2pt

$$c) r_n = \underline{k_n [M] [P]}$$

2pt

$$d) r_{opr} = \underline{k_t [P]^2}$$

2pt

$$e) r_{opr} = r_i$$

$$k_t [P]^2 = 2f k_d [I]$$

$$\rightarrow \underline{[P] = \sqrt{\frac{2f k_d [I]}{k_t}}}$$

2pt

$$f) r_n = k_n \cdot [M] \cdot \sqrt{\frac{2f k_d [I]}{k_t}}$$

$$= k_n \cdot [M] \cdot \sqrt{\frac{2f k_d [I]_0}{k_t}} e^{-k_d t}$$

} (b) ddka

2pt

$$g) \text{S\u00e4ttung} \quad [M] = [M]_0 : \quad p = \frac{[M]_0 - [M]_t}{[M]_0} = 0$$

$$\text{nach } [M]_t = 0 \quad p = \frac{[M]_0}{[M]_0} = \underline{1}$$

2pt

②

$$h) r_a = \frac{d[M]}{dt} = -k_a \cdot [M] \cdot \sqrt{\frac{2fk_d C I_2}{k_t q r}}$$

3pt

$$\rightarrow [M]_t = [M]_0 \cdot e^{-k_a \sqrt{\frac{2fk_d C I_2}{k_t q r}} t}$$

$$\rightarrow \rho = 1 - \frac{[M]_t}{[M]_0} = 1 - e^{-k_a \sqrt{\frac{2fk_d C I_2}{k_t q r}} t}$$

$$i) t \rightarrow \infty \quad e^{-\dots t} \rightarrow 0$$

$$\rightarrow \rho_\infty = 1 - e^{-\frac{2k_a \cdot \sqrt{\frac{2fk_d C I_2}{k_t \cdot k_{sp} r}}}{k_t \cdot k_{sp} r} t}$$

2pt

20pt

3.

(a) - (f) - katrs 0.5pt, kopā 1.6pt

Tabula atbildēm

savienojums	λ_{max}/nm	konjugētas saites	enerģijas līmeņu daudzums	n_{HOMO}	n_{LUMO}	sistēmas garums, pm	$n^2_{LUMO} - n^2_{HOMO}$	$\Delta E(teor)/J$	λ_{teor}/nm
hex-1-ene	215	1	2	1	2	290	3	2.15×10^{-18}	92
2,4-hexadiene	228	3	4	2	3	570	5	9.27×10^{-19}	214
1,3,5-hexatriene	272	5	6	3	4	850	7	5.85×10^{-19}	341
retinol	328	9	10	5	6	1410	11	3.33×10^{-19}	596
molekula X	482								

e) $\Delta E = E_{LUMO} - E_{HOMO}$

$$= \frac{h^2 n_{LUMO}^2}{8 m_e l^2} - \frac{h^2 n_{HOMO}^2}{8 m_e l^2}$$

$$= \frac{h^2}{8 m_e l^2} (n_{LUMO}^2 - n_{HOMO}^2)$$

tiem
precīzi
tikai
izteiksmē,
bet starp
pildz.

f) $\lambda = \frac{hc}{\Delta E} = \frac{8 c m_e l^2}{h (n_{LUMO}^2 - n_{HOMO}^2)}$

2pt

g) jebkurš sekotājs skaidrojums der

2pt

h) $y = 15x + 194$

$$482 = 15x + 194$$

2pt

$$x = 19.2 \approx 19 \text{ konj. Saistīti}$$

1.6pt
kopā